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dissociation enthalpy of  $H_2$ ,  $Cl_2$  and  $HCl$  are 434, 242 and 431  $kJmol^{-1}$  respectively. Enthalpy of formation of  $HCl$  is (1) 245  $kJmol^{-1}$  (2) 93  $kJmol^{-1}$  (3) -245  $kJmol^{-1}$  (4) -93  $kJmol^{-1}$  Sol. Answer (4) 61. Which of the following statements is not correct?

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energy of deuteron is 2.2 MeV and



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Solution 2  
that of  $4\text{He } 2$  is  $28\text{MeV}$ . If two deuterons are fused to form one  $4\text{He } 2$  then the energy released is:- (1)  $25.8\text{ MeV}$  (2)  $23.6\text{ MeV}$  (3)  $19.2\text{ MeV}$  (4)  $30.2\text{ MeV}$   
Q.11 In a radioactive material the activity at time  $t_1$  is  $R_1$  and at a later time  $t_2$ , it is  $R_2$ . If the decay

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