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L11-2 Lab 11 Atomic Spectra

OVERVIEW Light is an electromagnetic wave, its color is determined by its wavelength. The red light from a He-Ne laser has a wavelength of 632.8 nm.

Atomic Spectra Structure Lab Answers

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Lab Answer To observe and understand line emission spectra of atoms using gas-discharge tubes. To practice writing electron configurations for these (and other) elements. Electromagnetic radiation is energy in the form of waves. Waves are characterized by their wavelength (?) and frequency (?).

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5: Flame Tests and Atomic Spectra (Experiment) - Chemistry ...

Atomic Emission Line Spectra. In a gas discharge tube energy is added to a gas which is adsorbed by electrons. These excited electrons enter high energy orbitals and then fall back to

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their lower energy and can give off a photon of light of a specific wavelength that can be seen by a diffraction grating or prism. These type of spectra are called line spectra (in contrast to continuous spectra ...

6.3: Atomic Line Spectra and Niels

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