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a. equals the sum of the mass numbers of the products. b. does not change after the decay. c. is the same as the atomic number. d.

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cannot be determined. a. equals the sum of the mass numbers of the products. When a nuclear undergoes nuclear decay by gamma rays, the atomic number of the element. a. remains the same.

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radon-222.

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Chapter 10 Nuclear Chemistry Summary 10.1 Radioactivity •

Radioactivity is the process in which an unstable atomic nucleus emits charged particles and energy. • Any atom containing an unstable nucleus is called a radioactive isotope, or radioisotope for short. During nuclear decay, atoms of one element can change into atoms of

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Terms in this set (26) Radioactive Decay. the disintegration of an unstable atomic nucleus into one or more different nuclides, accompanied by the emission of radiation, the nuclear capture or ejection of electrons, or fission.

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