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17 Practice Problems In your notebook, solve the following problems. SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK Use the three-step problem-solving approach you learned in Chapter many kilojoules of energy are in a donut that contains 200.0 Calories? 2. What is the specific heat of a substance that has a mass of 25.0 g and requires

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Ch 17 Thermochemistry Practice Test Specific Heat Problems 1) How much heat must be absorbed by 375 grams of water to raise its temperature by 25° C? 2) What mass of water can be heated from 25.0° C to 50.0° C by the addition of 2825 J? 3) What is the final temperature when 625 grams of water at 75.0° C loses  $7.96 \times 10^4$  J? Chapter 17.

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A 12.2 g sample of an unknown metal sample was heated to 98.6°C and then put into a calorimeter containing 25.0 ml of water at 22.3°C. After mixing, the temperature of the water and metal increased to a maximum of 28.4°C after 35 seconds.

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