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Considering the reaction represented by the (unbalanced) equation determine the number of moles of $\text{NH}_3(\text{g})$ that can be produced from the following: a. 0.20 mol $\text{N}_2(\text{g})$ reacts completely with $\text{H}_2(\text{g})$. b. 0.30 mol $\text{H}_2(\text{g})$ reacts completely with $\text{N}_2(\text{g})$. 2. Considering the reaction

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represented by the (unbalanced) equation

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Levels (GENERIC) and the third section is GED Core Content Areas (GCD). These are the last four of eight questions, the students will be required to answer.

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1 There are four stable isotopes of iron with mass numbers 54, 56, 57, and 58

There are two radioactive

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Chemistry Answers

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