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Periodicity in

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11 Chemistry

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~~Stoichiometry~~

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of physical
units (SI Units
) based on the
meter, kilogram,

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second, ampere,
kelvin, candela,
and mole,

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to indicate
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Review Answer

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questions in

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question). Show

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The law of conservation of mass follows from the concept that _____. 2.

If 4 g of element A combine with 10 g of element B, then 12 g of element A combine with _____ g of element B. 3.

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five principles
of the atomic...
law of
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mass. law of

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Review
proportions. law
of definite

proportions. 1.

all matter is

composed of

extremely small

particles

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nor destroyed in

a chemical

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and takes up

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or H₂). Elements

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two or more

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Chapter 2 56

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**Discovering
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Review Skills.

The presentation of information in this chapter assumes that you can already perform the tasks listed

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Review
below. You can
test your
readiness to
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answering the
Review .

Questions at the
end of the
chapter. This
might also be a
good time to
read the Chapter
. Objectives,
which precede

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**Chapter 3 – An
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up of motionless
particles.

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of definite proportions. compounds are made of the same elements in the same proportions. law of conservation of mass. mass cannot be created or destroyed. law of multiple proportions.

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measure of the

ability of an

atom in a

chemical

compound to

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Review
electrons from another atom in the compound is called _____.

17. The energy required to remove one electron from an atom is called its _____.

18.

Key topics:

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nomenclature,
Lavoisiers list
of elements,
sulfur,
diamonds,
graphite, coal,
medieval metals,
platinum, zinc,
cobalt, nickel,
manganese
molybdenum,
tungsten, gases
in the

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pressure and

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and Mass Number,

Isotopes, Mass
Spectroscopy 4.

Electron Energy

Levels 5. The

Periodic Table

and Periodic

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Symbols, Radii,

Ionization

Energy, Electron

Affinity,

Oxidation States

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B. BONDING 1.
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on States,
Structures, and
Properties of
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4. Geometry of
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and Coordination
Complexes 5.

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3. Oxidation and
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Balancing

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of Reaction 2.
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3. Temperature

Changes and

Effect on Rate

4. Activation

Energy 5.

Mechanism of a

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Thermodynamics

4. Change in

Free Energy

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Vertical, and

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Relationships in

the Periodic

Table 2.

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Elements and
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VI FORMULAS AND

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and thereby

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exam is divided
into two

sections: 1)

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your ability to

recall and

understand a

broad range of

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concepts and

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constitutes 45%
of the final
grade and you
are allowed 90
minutes for this
portion of the
exam.

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bring an electronic hand-held calculator for use on the 40-minute free-response section. Essay and chemical-reaction questions comprise the last 50 minutes of the test, during which

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section of the
exam is scored
by crediting
each correct
answer with one

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point, and deducting only partial credit (one-fourth of a point) for each incorrect answer. Omitted questions receive neither a credit nor a deduction. The essay section is scored by a group of more

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than 1,000 college and high school educators

familiar with

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evaluate the

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essays

accordingly. The

grades given for

the essays are

combined with

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the results of
the multiple-
choice section,

and the total

raw score is

then converted

to the

program''s five-

point scale: 5 -

Extremely well

qualified 4 -

Well qualified 3

- Qualified 2 -

Possibly

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need to help get
the score you
want."--

O Level

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Multiple Choice

Questions and

Answers (MCQs) :

Quizzes &

Practice Tests

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with Answer Key
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tests for

competitive

exams to solve

899 MCQs. "O

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analytical study

for self-

assessment,

career tests.

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Questions and

Answers (MCQs)

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topics: Acids

and bases,

chemical bonding

and structure,

chemical

formulae and

equations,

electricity,

electricity and

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chemicals,
elements,
compounds,
mixtures, energy
from chemicals,
experimental
chemistry,
methods of
purification,
particles of
matter, redox
reactions, salts
and
identification

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of ions and gases, speed of reaction, and structure of atom to enhance teaching and learning. 0
Level Chemistry
Quiz Questions
and Answers also
covers the
syllabus of many
competitive
papers for

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Chemistry

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admission exams
of different
universities

from chemistry
textbooks on

chapters: Acids
and Bases

Multiple Choice
Questions: 123

MCQs Chemical
Bonding and

Structure

Multiple Choice
Questions: 75

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MCQs Chemical
Formulae and
Equations

Multiple Choice
Questions: 167

MCQs Electricity
Multiple Choice
Questions: 107

MCQs Electricity
and Chemicals
Multiple Choice
Questions: 10

MCQs Elements,
Compounds and

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Chemistry

Mixtures 3 Test

Multiple Choice

Questions: 39

MCQs Energy from
Chemicals

Multiple Choice

Questions: 41

MCQs

Experimental

Chemistry

Multiple Choice

Questions: 18

MCQs Methods of
Purification

Acces PDF

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Chapter 3 Test

Questions: 84

MCQs Particles

of Matter

Multiple Choice

Questions: 45

MCQs Redox

Reactions

Multiple Choice

Questions: 42

MCQs Salts and

Identification

of Ions and

Gases Multiple

Acces PDF
Chemistry
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Review

Questions: 61

MCQs Speed of
Reaction

Multiple Choice
Questions: 35

MCQs Structure
of Atom Multiple
Choice

Questions: 52

MCQs The chapter
"Acids and Bases
MCQs" covers
topics of acid

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Chemistry

Chapter 3 Test

rain, acidity
needs water,

acidity or

alkalinity,

acids properties

and reactions,

amphoteric

oxides, basic

acidic neutral

and amphoteric,

chemical

formulas,

chemical

reactions,

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Chemistry
Chapter 3 Test

Review
reactions,
college
chemistry,
mineral acids,
general
properties,
neutralization,
ordinary level
chemistry,
organic acid, pH
scale, acid and
alkali,
properties,

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bases and reactions, strong and weak acids, and universal indicators. The chapter "Chemical Bonding and Structure MCQs" covers topics of ions and ionic bonds, molecules and covalent

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bonds,
evaporation,
ionic and
covalent
substances,
ionic compounds,
crystal
lattices,
molecules and
macromolecules,
organic
solvents,
polarization,
and transfer of

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"Chemical
Formulae and
Equations MCQs"
covers topics of
chemical
formulas,
chemical
equations,
atomic mass,
ionic equations,
chemical
reactions,

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Chemistry
Chapter 3 Test
Review

symbols,
mixtures and
compounds, molar
mass, percent
composition of
elements,
reactants,
relative
molecular mass,
valency and
chemical
formula, and
valency table.

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"Electricity
MCQs" covers

topics of
chemical to
electrical
energy,
applications of
electrolysis,
reactions,
conductors and
non-conductors,
dry cells,
electrical

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Review
devices, circuit
symbols,
electrolytes, no

n-electrolytes,
organic

solvents,

polarization,

and valence

electrons. The

chapter

"Electricity and
Chemicals MCQs"

covers topics of
chemical to

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Chemistry
Chapter 3 Test
Review

energy, dry
cells,

electrolyte, non-
electrolyte, and
polarization.

The chapter

"Elements,

Compounds and

Mixtures MCQs"

covers topics of

elements,

compounds,

mixtures,

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Review
molecules,
atoms, and
symbols for
elements.

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with few
additional

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practice with 3

full-length

practice tests

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exam experience

with a timed

test option

Deepen your

understanding

with detailed

answer

explanations and

expert advice

Gain confidence

with automated

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scoring to check
your learning
progress

Master the SAT
II Chemistry
Subject Test and
score higher...
Our test experts
show you the
right way to
prepare for this
important
college exam.

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REA's SAT II Test

Chapter 3
Review
Chemistry test
prep covers all
chemistry topics
to appear on the
actual exam
including in-
depth coverage
of the laws of
chemistry,
properties of
solids, gases
and liquids,
chemical

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Chapter 3 Test

Review
reactions, and
more. The book
features 6 full-

length practice
SAT II Chemistry

exams. Each

practice exam

question is

fully explained

to help you

better

understand the

subject

material. Use

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Periodic Table
Review

of Elements for
speedy look-up
of the

properties of
each element.

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study with REA's
proven test-
taking

strategies,
powerhouse

drills and study

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schedule that
get you ready
for test day.

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review of every
chemistry topic
to appear on the
SAT II subject
test - Flexible
study schedule
tailored to your
needs - Packed
with proven test

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tips, strategies
and advice to

help you master

the test - 6

full-length

practice SAT II

Chemistry

Subject tests.

Each test

question is

answered in

complete detail

with easy-to-

follow, easy-to-

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Chapter 3 Test
Review

explanations. -
The book's handy
Periodic Table
of Elements
allows for quick
answers on the
elements
appearing on the
exam

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(The Mole
Concept)

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Formulas Mole

Concept

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and Formula

Weight

Equivalent

Weight Chemical

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Composition Stoichiometry/Weight
and Volume

Calculations

Balancing

Chemical

Equations

Calculations

Based on

Chemical

Equations Limiti

ng-Reactant

Calculations

Solids Phase

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Equilibrium

Properties of

Liquids Density

Colligative

Properties of

Solutions

Raoult's Law and

Vapor Pressure

Osmotic Pressure

Solution

Chemistry

Concentration

Units

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Equilibrium The
Law of Mass

Action Kinetics
and Equilibrium

Le Chatelier's
Principle and

Chemical

Equilibrium Acid-
Base Equilibria

Definitions of

Acids and Bases

Ionization of

Water, pH

Dissociation of

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Weak

Electrolytes

Dissociation of

Polyprotic Acids

Buffers

Hydrolysis

Thermodynamics I

Bond Energies

Some Commonly

Used Terms in

Thermodynamics

The First Law of

Thermodynamics

Enthalpy Hess's

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Summation

Standard States

Heat of

Vaporization and

Heat of Fusion

Thermodynamics

II Entropy The

Second Law of

Thermodynamics

Standard

Entropies and

Free Energies

Electrochemistry

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Chapter 3 and
Oxidation and

Reduction

Electrolytic

Cells Non-

Standard-State

Cell Potentials

Atomic Theory

Atomic Weight

Types of Bonds

Periodic Trends

Electronegativit

y Quantum

Chemistry Basic

Electron Charges

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Components of
Atomic Structure

The Wave

Mechanical Model

Subshells and

Electron

Configuration

Double and

Triple Bonds

Organic

Chemistry:

Nomenclature and

Structure

Alkanes Alkenes

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Chemistry

Chapter 3 Test

Alkyl Halides

Cyclic

Hydrocarbons

Aromatic

Hydrocarbons

Aryl Halides

Ethers and

Epoxides

Alcohols and

Glycols

Carboxylic Acids

Carboxylic Acid

Derivatives

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Esters Amides
Arenes Aldehydes
and Ketones

Amines Phenols
and Quinones

Structural

Isomerism SIX

PRACTICE EXAMS

"Practice Test 1

" Answer Key

Detailed

Explanations of

Answers

"Practice Test 2

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" Answer Key Test

Detailed

Explanations of

Answers

"Practice Test

3" Answer Key

Detailed

Explanations of

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"Practice Test 4

" Answer Key

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Explanations of

Answers

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Detailed

Explanations of

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"Practice Test 6

" Answer Key

Detailed

Explanations of

Answers THE

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Education

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faculties of

prestigious high

schools,

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throughout the

United States

and Canada.

CHAPTER 1 -

ABOUT THE SAT

II: CHEMISTRY

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SUBJECT TEST

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This book provides you with an accurate and complete representation of the SAT II: Chemistry Subject Test.

Inside you will find a complete course review designed to

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provide you with the information and strategies needed to do well on the exam, as well as six practice tests based on the actual exam. The practice tests contain every type of question that you can expect

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Following each

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find an answer

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detailed

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test material.

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Test and What Is
It Used For?

Students

planning to

attend college

take the SAT II:

Chemistry

Subject Test for

one of two

reasons: (1)

Because it is an

admission

requirement of

the college or

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which they are
applying; "OR"

(2) To
demonstrate
proficiency in
Chemistry. The
SAT II:
Chemistry exam
is designed for
students who
have taken one
year of college
preparatory

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Who Administers The Test? The SAT II: Chemistry Subject Test is developed by the College Board and administered by Educational Testing Service (ETS). The test development process involves the assistance

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Review
of educators throughout the country, and is designed and implemented to ensure that the content and difficulty level of the test are appropriate.

When Should the SAT II:

Chemistry be Taken? If you

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Review
are applying to
a college that
requires Subject

Test scores as
part of the
admissions

process, you
should take the
SAT II:

Chemistry

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toward the end
of your junior
year or at the

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beginning of
your senior
year. If your
scores are being
used only for
placement
purposes, you
may be able to
take the test in
the spring of
your senior
year. For more
information, be
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the colleges to
which you are
applying. When

and Where is the
Test Given? The
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Subject Test is
administered
five times a
year at many
locations

throughout the
country; mostly

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high schools. To receive information on upcoming administrations of the exam, consult the publication *Taking the SAT II: Subject Tests*, which may be obtained from your guidance counselor or by

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SAT Program P.O.

Box 6200

Princeton, NJ

08541-6200

Phone: (609)

771-7600

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II: Chemistry.

Consult the

publication

Taking the SAT

II: Subject

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information on

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structure.

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Subject Test is

designed to test
knowledge that

has been

acquired

throughout your
education.

Therefore, the

best way to

prepare for the

exam is to

refresh yourself

by thoroughly

studying our

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Review
review material
and taking the
sample tests

provided in this
book. They will
familiarize you
with the types
of questions,
directions, and
format of the
SAT II:

Chemistry

Subject Test. To
begin your

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Review
studies, read
over the review
and the

suggestions for
test-taking,
take one of the
practice tests
to determine
your area(s) of
weakness, and
then restudy the
review material,
focusing on your
specific problem

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areas. The course review includes the information you need to know when taking the exam. Be sure to take the remaining practice tests to further test yourself and become familiar with the format

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Chemistry
of the SAT II:
Chemistry
Review
Subject Test.

When Should I
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early to start
studying for the
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Chemistry test.
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begin, the more
time you will
have to sharpen

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your skills. Do not

procrastinate!

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to study, since

it does not

allow you the

time needed to

learn the test

material. The

sooner you learn

the format of

the exam, the

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Review
more comfortable
you will be when
you take the

exam. FORMAT OF
THE SAT II:

CHEMISTRY The
SAT II:

Chemistry is a
one-hour exam
consisting of 85
multiple-choice
questions. The
first part of
the exam

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Chapter 3 Test

Review
consists of classification questions. This question type presents a list of statements or questions that you must match up with a group of choices lettered (A) through (E). Each choice may be used once,

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Chapter 3 Test

Review

more than once,
or not at all.
The exam then
shifts to
relationship
analysis
questions which
you will answer
in a specially
numbered section
of your answer
sheet. You will
have to
determine if

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Chapter 3 Test

Review

each of two statements is true or false and if the second statement is a correct explanation of the first. The last section is composed strictly of multiple-choice questions with choices lettered

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(A) through (E).

Material Tested

The following

chart summarizes

the distribution

of topics

covered on the

SAT II:

Chemistry

Subject Test.

Topic /

Percentage /

Number of

Questions Atomic

Acces PDF
Chemistry
& Molecular Test
Structure / 25%
/ 21 questions
States of Matter
/ 15% / 13
questions
Reaction Types /
14% / 12
questions
Stoichiometry /
12% / 10
questions
Equilibrium &
Reaction Times /

Acces PDF

Chemistry

Chapter 3 Test
7% / 6 questions

Thermodynamics /

6% / 5 questions

Descriptive

Chemistry / 13%

/ 11 questions

Laboratory / 8%

/ 7 questions

The questions on

the SAT II:

Chemistry are

also grouped

into three

larger

Acces PDF

Chemistry

Chapter 3 Test

Review
according to how they test your understanding of the subject material.

Category /

Definition /

Approximate

Percentage of

Test 1) Factual

Recall /

Demonstrating a

knowledge and

Acces PDF

Chemistry

Chapter 3 Test

Review

Understanding of
important

concepts and
specific

information /

20% 2)

Application /

Taking a

specific

principle and

applying it to a

practical

situation / 45%

3) Integration /

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Chapter 3 Test

Inferring
information and
drawing

conclusions from
particular
relationships /

35% STUDYING FOR
THE SAT II:

CHEMISTRY It is
very important
to choose the
time and place
for studying
that works best

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Review

for you. Some students may set aside a certain number of hours every morning to study, while others may choose to study at night before going to sleep. Other students may study during the day, while waiting on line,

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or even while eating lunch. Only you can determine when and where your study time will be most effective. Be consistent and use your time wisely. Work out a study routine and stick to it! When you take

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Chapter 3 Test

Review
the practice tests, try to make your

testing

conditions as

much like the

actual test as

possible. Turn

your television

and radio off,

and sit down at

a quiet desk or

table free from

distraction.

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Review
Make sure to clock yourself with a timer. As you complete each practice test, score it and thoroughly review the explanations to the questions you answered incorrectly; however, do not review too much

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Review
at any one time.
Concentrate on
one problem area

at a time by
reviewing the
questions and
explanations,
and by studying
our review until
you are

confident you
completely
understand the
material. Keep

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track of your scores. By doing so, you will be able to gauge your progress and discover general weaknesses in particular sections. You should carefully study the reviews that cover your areas

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Review

of difficulty,
as this will
build your
skills in those
areas. TEST

TAKING TIPS

Although you may
be unfamiliar
with

standardized
tests such as
the SAT II:

Chemistry

Subject Test,

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Chapter 3 Test

Review
there are many
ways to acquaint
yourself with

this type of
examination and
help alleviate
your test-taking
anxieties.

Become
comfortable with
the format of
the exam. When
you are
practicing to

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Chapter 3 Test II:

Chemistry

Subject Test,

simulate the

conditions under

which you will

be taking the

actual test.

Stay calm and

pace yourself.

After simulating

the test only a

couple of times,

you will boost

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Chapter 3 Test Review

your chances of doing well, and you will be able to sit down for the actual exam with much more confidence. Know the directions and format for each section of the test.

Familiarizing yourself with the directions

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Review

and format of the exam will not only save you time, but will also ensure that you are familiar enough with the SAT II: Chemistry Subject Test to avoid nervousness (and the mistakes caused by being

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Review
Do your scratchwork
in the margins

of the test

booklet. You

will not be

given scrap

paper during the

exam, and you

may not perform

scratchwork on

your answer

sheet. Space is

provided in your

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Chapter 3 Test

Review
test booklet to
do any necessary
work or draw

diagrams. If you
are unsure of an
answer, guess.

However, if you
do guess - guess
wisely. Use the
process of
elimination by
going through
each answer to a
question and

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ruling out as many of the answer choices as possible. By eliminating three answer choices, you give yourself a fifty-fifty chance of answering correctly since there will only be two choices

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Review
left from which
to make your
guess. Mark your
answers in the
appropriate
spaces on the
answer sheet.

Fill in the oval
that corresponds
to your answer
darkly,
completely, and
neatly. You can
change your

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Review

answer, but remember to completely erase your old answer. Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly. When you have finished working on a section,

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Review
you may want to
go back and
check to make

sure your

answers

correspond to

the correct

questions.

Marking one

answer in the

wrong space will

throw off the

rest of your

test, whether it

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Chapter 3 Test

is graded by
machine or by
Review
hand. You don't

have to answer
every question.

You are not
penalized if you
do not answer
every question.

The only penalty
results from
answering a
question

incorrectly. Try

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to use the guessing strategy, but if you are truly stumped by a question, remember that you do not have to answer it. Work quickly and steadily. You have a limited amount of time to work on each

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Review

section, so you need to work quickly and steadily. Avoid focusing on one problem for too long. Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost

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on the day of the test. On the night before the test, gather together the materials you will need the next day: - Your admission ticket - Two forms of identification (e.g., driver's license, student identification

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Review

alien
registration

card) - Two No.

2 pencils with

erasers -

Directions to

the test center

- A watch (if

you wish) but

not one that

makes noise, as

it may disturb

other test-

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Chapter 3 Test

takers On the day of the test, you should wake up early (after a good night's rest) and have breakfast. Dress comfortably, so that you are not distracted by being too hot or too cold while taking the test. Also, plan to

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Review

arrive at the test center early. This will allow you to collect your thoughts and relax before the test, and will also spare you the stress of being late. If you arrive after the test begins, you will not be

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Chapter 3 Test
Review
admitted to the
test center and
you will not

receive a
refund. During
the Test When
you arrive at
the test center,
try to find a
seat where you
feel most
comfortable.

Follow all the
rules and

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Review
instructions
given by the
test supervisor.

If you do not,
you risk being
dismissed from
the test and
having your
scores canceled.

Once all the
test materials
are passed out,
the test
instructor will

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give you directions for filling out your answer sheet.

Fill this sheet out carefully since this information will appear on your score report.

After the Test

When you have completed the SAT II:

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Subject Test, Review

you may hand in
your test

materials and

leave. Then, go
home and relax!

When Will I

Receive My Score
Report and What

Will It Look

Like? You should
receive your

score report

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about five weeks after you take the test. This report will include your scores, percentile ranks, and interpretive information.

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and the highest-
yield questions

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most-tested
concepts plus
more questions

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MCAT KNOWLEDGE:

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team has spent

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every document

related to the

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with our expert
psychometricians
, the Kaplan
team is able to
ensure the
accuracy and
realism of our
practice
materials.

THOROUGH SUBJECT
REVIEW: Written
by top-rated,
award-winning
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EXPANDED CONTENT
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While the MCAT has continued to develop, this book has been updated

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continuously to
match the AAMC's
guidelines

precisely—no

more worrying if

your prep is

comprehensive!

MORE PRACTICE

THAN THE

COMPETITION:

With questions

throughout the

book and access

to one practice

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test, Kaplan's
MCAT General

Review
Chemistry Review

has more

practice than

any other MCAT

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Chemistry book

on the market.

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COMPANION:

Access to online

resources to

augment content

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studying, including one practice test.

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and diagrams
from the pages
of Scientific

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turns even the

most intangible,

complex science

into easy-to-

visualize

concepts.

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