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Matter And  
Change Chapter  
Assessments**

**Chapter**

**Assessments**

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Matter and Change

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**part 1 ( Chem 1 H)**

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*Chemistry \"The Three*

*States of Matter\" The*

*Creation of Chemistry -*

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*The Fundamental Laws:*

*Crash Course*

*Chemistry #3 Matter*

*and Change Chapter 1*

*Introduction: Matter and*

*Measurement States of*

*matter | States of matter*

*and intermolecular*

*forces | Chemistry |*

*Khan Academy Grade 9*

*Chemistry Lesson 1 -*

*Matter and the Particle*

*Theory What Is Matter?*

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and Change Part 1~~

Particulate nature of  
matter - Part-1 -

**Chemistry Changes of  
States of Matter  
Chemical and Physical  
Changes**

---

States of Matter : Solid  
Liquid Gas *10 Amazing  
Experiments with Water*

---

Pure Substances and  
*Page 6/33*

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Chemistry

Mixtures! And

(Classification of  
Matter)Physical and  
Chemical Changes

Understanding Atoms,  
elements, and molecules  
Part #1 (9min)

Basic Chemistry Series -  
1.2 What is matter? -  
HD Properties of Matter  
Physical and Chemical  
Changes: Chemistry for  
Kids - FreeSchool Phase  
Changes States of

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~~Matter Solids, Liquids,  
Gases \u0026 Plasma  
Change Chapter  
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8th/Chapter 1 /Matter

~~Classifying Matter With  
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Chemicals Ozone Layer,

atmosphere, ozone

formation,

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chlorofluorocarbons,  
CFC's 1.2 Chemistry  
and Matter Chemistry  
Central Science

**Chapter Summaries –  
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Chapter 1 Section 2

MATTER and Its

Properties Matter –

Anything that has

MASS and VOLUME.

ATOMS and

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**MOLECULES** the basic building blocks of matter. **ATOM**-- the smallest unit of an element that still maintains the properties that element.

**ELEMENT**-- a pure substance made up of only one type of atom.

## **MATTER and CHANGE Chapter 1 Section 1**

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Chemistry: Matter and  
Change Chapter 6 Study  
Guide for Content  
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CHAPTER Class  
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Section 6.3 Periodic  
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textbook, read about  
atomic radius and ionic  
radius. Circle the letter  
of the choice that best  
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or answers the question.

1. Atomic radii cannot  
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SOLUTIONS  
MANUAL

Matter—Properties and C  
hangesMatter—Propertie  
s and Changes Solutions

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Manual Chemistry:

Matter and Change •  
Chapter 3 35 Section 3.1

Properties of Matter

pages 70–75 Problem-

Solving Lab 1. Explain

why the flow of a

compressed gas must be

controlled for practical

and safe use.

**Matter—Properties and  
Changes**  
**Matter—Prope  
rties and Changes**



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Chapter 1 VIDEODISC

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Periodic Law Section  
6.1 Development of the  
Modern Periodic Table  
1. octaves 2. eight 3.  
nine 4. accepted 5.

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atomic mass 7. elements  
8. atomic number 9.  
Henry Moseley 10.  
protons 11. periodic law  
12. properties 13. 14 ...

## **Ch 6 Study Guide answers**

Date Class Chapter  
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Chapter 5 25 Electrons  
in Atoms Reviewing

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Vocabulary Match the definition in Column A with the term in Column B. Column A Column B

1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element
2. The minimum amount of energy that can be lost or gained by ...

## **Chapter Assessment**

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(that is, energy

produced or absor... A

system that can

exchange neither energy

nor matter with its...

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extensive property.

intensive property.

extensive properties

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property (examples) a  
property that depends on  
the amount of matter in  
a sample. a property that  
depends on the type of  
matter in a sample not....  
mass, volume, length,  
weight.

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number of moles of

known substance C H C

H C H O O O CO CO

CO H O H O H O

number of moles of

unknown substance O

CO H O C H CO H O C

H O H O C H O CO use

mole ratio 5 mol O 2 1

mol C 3 H 8 3 mol CO 2

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1 mol C 3 H 8 4 mol H

2 O 1 mol C 3 H 8 1

mol C 3 H 8 5 mol O 2

## **MATH SKILLS**

## **TRANSPARENCY**

## **MASTER 16**

A “chemical change” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one or

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more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of  
Compounds Quite  
different properties than  
their component  
elements.



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