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This collection of ten chemistry test questions deals with the concepts introduced with the ideal gas laws. Useful information: At STP : pressure = 1 atm = 700 mm Hg, temperature = 0 °C = 273 K At STP: 1 mole of gas occupies 22.4 L R = ideal gas constant = 0.0821 L·atm/mol·K = 8.3145 J/mol·K Answers appear at the end of the test.

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