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ANSWER KEY 1 Modern chemistry answer key chapter 2. a 2. b 3. c 4. c 5. a 6. b 7 Modern chemistry answer key chapter 2. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a

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solution is acidic or basic.

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CHAPTER 2 REVIEW Measurements and Calculations

SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches

### 2 Measurements and Calculations

Modern Chemistry 105 Chapter Test Name Class Date

Chapter Test A, continued Use this figure to answer

questions 7 and 8. \_\_\_\_\_ 7. A solution containing 35 g of  $\text{Li}_2\text{SO}_4$  dissolved in 100 g of water is heated from  $10^\circ\text{C}$  to

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90°C. According to information in the figure, this temperature change would result in a. an additional 5 g of  $\text{Li}_2\text{SO}_4$  in ...

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2O shows that 2 molecules (made of 4 atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses of each of the reactants, hydrogen and oxygen. What parts of Dalton's atomic theory are illustrated by this reaction?

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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill  
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PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided. 25. 26.

MODERN CHEMISTRY The atomic number of nickel-60 is

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28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11

CHAPTER 3 TEST

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW  
SHORT ANSWER Answer the following questions in the  
space provided. 1. Classify each of the following as a

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homogeneous or heterogeneous substance. homogeneous a. sugar homogeneous d. plastic wrap homogeneous b. iron filings heterogeneous e. cement sidewalk heterogeneous c. granola bar 2. For each type of investigation, select the most appropriate ...

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### Quizlet

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### 10 States of Matter - Ms. Agostine's Chemistry Page

Algebra 1: Common Core (15th Edition) Charles, Randall I.  
Publisher Prentice Hall ISBN 978-0-13328-114-9

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AbeBooks Modern Chemistry 179 Organic Chemistry  
CHAPTER 22 REVIEW Organic Chemistry SECTION 1  
SHORT ANSWER Answer the following questions in the  
space provided. 1. Name two types of carbon-containing  
molecules that are not organic. \_\_\_\_\_ 2. \_\_\_\_\_ Carbon  
atoms form bonds readily with atoms of ...

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