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(a) has mobile electrons in  
the crystal c covalent  
molecular crystal (b) is  
hard, brittle, and  
nonconducting a metallic  
crystal (c) typically has  
the lowest melting point of

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CHAPTER 10 REVIEW States of

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Matter SECTION 4 SHORT

ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two

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opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

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Chapter 10. States of  
Matter. Kinetic-Molecular  
Theory The kinetic-molecular  
theory of matter is based on

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the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties. Kinetic-Molecular Theory of Gases 1- Gases consist of large numbers of tiny particles

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that are far apart relative to their size. 2- Collisions between gas particles and between particles and their container walls are elastic.

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States of Matter*

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REVIEW States of Matter

SECTION 1 SHORT ANSWER

Answer the following questions in the space provided.

1. Identify whether the descriptions below describe an ideal gas

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or a real gas. ideal gas  
Review Answers

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*CHAPTER 10 REVIEW SECTION 1.*

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