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Chapter 18 - An Introduction to Chemistry: Nuclear Chemistry (See Section 4.1 for a review of the different forms of radiant energy.)

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Chapter 18 covers leaving groups that are directly attached to double-bonded sp²carbons. These molecules don't do most of the regular alkyl halide chemistry from Ch. 9 (SN1/ SN2/E1), but they can do E2 under the right circumstances. (This is in strong contrast to Ch. 17.

~~Loudon Chapter 18 Review: Vinyl/Aryl Reactivity~~

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Iron (Fe) is the strongest reducing agent because it is always oxidized. It reduces all of the other metals. Gold (Au) is the weakest reducing agent because it is never oxidized. Using the voltages of the other metals with Pb.

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