

Note Taking Guide Episode 303

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Note Taking Guide: Episode 303 Name Aditi Acharya Bohr's Energy Levels | Electrons in certain energy levels | Lower energy levels: closer to nucleus | Higher energy levels: further from nucleus | Ground State: electron in lowest energy level possible Excited Atom | Atom has absorbed energy | Excited state is unstable | Atom soon emits same amount of energy absorbed | energy seen as visible light Wave Description of Light Wavelength (λ): distance between corresponding points ...

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lowest energy level possible Excited Atom □ Atom has absorbed

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